



Pergamon

Tetrahedron Letters 41 (2000) 2659–2662

TETRAHEDRON
LETTERS

Optically active *trans*-4-(*tert*-butyldimethylsiloxymethyl)-5-(*tert*-butyldimethylsiloxy)-2-cyclohexenone as a useful chiral building block for preparation of substituted cyclohexane rings: synthesis and its highly stereoselective reaction with $\text{RCu}(\text{CN})\text{Li}$

Takeshi Hanazawa, Masakazu Koiwa, Georges P.-J. Hareau and Fumie Sato *

Department of Biomolecular Engineering, Tokyo Institute of Technology, 4259 Nagatsuta-cho, Midori-ku, Yokohama, Kanagawa 226-8501, Japan

Received 17 January 2000; revised 3 February 2000; accepted 4 February 2000

Abstract

Optically active *trans*-4-(*tert*-butyldimethylsiloxymethyl)-5-(*tert*-butyldimethylsiloxy)-2-cyclohexenone (**2**) has been synthesized in 25% overall yield starting from easily available 1,4-bis(benzyloxy)-2,3-epoxy butane (**3**). The enone **2** reacts with excellent stereoselectivity with $\text{RCu}(\text{CN})\text{Li}$ thus working as an efficient chiral building block for preparation of substituted cyclohexane compounds. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: asymmetric synthesis; copper; copper compounds; cyclohexenones; cyclohexanes.

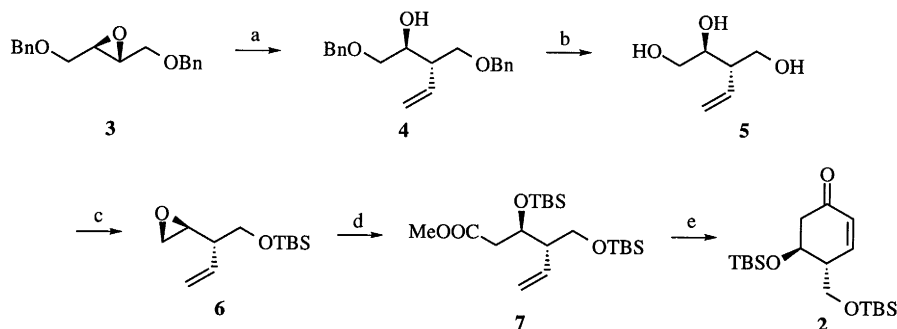
Many biologically important compounds have a chiral cyclohexane ring in their structure as the main or a subunit. One attractive method to synthesize these skeletons involves the use of naturally occurring or man-made non-racemic 2-cyclohexenone compounds as the starting material, taking advantage of their versatile reactivity. Recently, we have developed an efficient and practical method for preparation of optically active 5-(*tert*-butyldimethylsiloxy)-2-cyclohexenone (**1**) and have shown that it can be effectively used as a chiral building block for synthesizing a variety of substituted 2-cyclohexenone and cyclohexanone derivatives.¹

However, it is not an easy task to prepare 2-cyclohexenone or cyclohexanone derivatives having a substituent at the 4-position starting from **1**. Optically active 5-(*tert*-butyldimethylsiloxy)-2-cyclohexenone having a siloxymethyl group at the 4-position seems to be an attractive chiral building block for the preparation of a variety of these types of compounds, since the siloxymethyl group can be potentially used as a handle for further functionalization. We report here the synthesis of 4-(*tert*-butyldimethylsiloxymethyl)-5-(*tert*-butyldimethylsiloxy)-2-cyclohexenone with *trans* configuration, **2**, starting from 1,4-bis(benzyloxy)-2,3-epoxy butane (**3**), both enantiomers of which can be readily pre-

* Corresponding author. Tel: +0081 45 924 5787; fax: +0081 45 924 5826; e-mail: fsato@bio.titech.ac.jp (F. Sato)

pared from diethyl D- or L-tartrate, respectively, according to the protocol of Nicolaou et al.,² and its highly diastereoselective conjugate addition reaction with an organocopper compound.

The preparation of **2** from **3** was carried out according to the procedure shown in Scheme 1. Reaction of (*R,R*)-**3** with vinylmagnesium bromide in the presence of CuI gave the vinylalcohol **4** (94% yield),³ which in turn was converted to the triol **5** under the Birch reduction conditions (88% yield). From **5**, epoxide **6** was prepared via a four-step sequence according to the procedure developed by Sharpless in 64% overall yield.⁴ Compound **6** was then converted to the ester **7** in 62% overall yield by conventional reaction sequences which involve ring opening with the CN anion using Et₂AlCN, protection of the resulting hydroxy group as a TBS-ether, and the following conversion of the CN moiety to a methylester group. The ester **7** could then be converted to the expected **2** in 72% overall yield according to the procedure used previously for the synthesis of **1**, i.e., the intramolecular nucleophilic acyl substitution reaction mediated by a Ti(*O-i-Pr*)₄/2*i-Pr*MgCl reagent⁵ and FeCl₃-induced ring enlargement reaction.⁶ In conclusion, the enone **2** was prepared from **3** in 25% overall yield.



Scheme 1. (a) CuI, CH₂=CHMgBr, Et₂O; (b) Li/NH₃, THF; (c) (i) MeC(OMe)₃, cat. PPTS, CH₂Cl₂, (ii) CH₃COBr, CH₂Cl₂, (iii) K₂CO₃, MeOH, (iv) TBSCl, imidazole, DMF; (d) (i) Et₂AlCN, THF, (ii) TBSCl, imidazole, DMF, (iii) DIBAL-H, C₆H₁₄, (iv) NaClO₂, NaH₂PO₄·2H₂O, H₂O, ^tBuOH, 2-methyl-2-butene, (v) MeI, K₂CO₃, acetone; (e) (i) Ti(*O-i-Pr*)₄/2*i-Pr*MgCl, Et₂O, (ii) FeCl₃, pyr, DMF, (iii) AcONa, MeOH

With **2** in hand, our next concern was the diastereoselectivity of the conjugated addition of organocopper compounds to it. It had been reported that 1,4-addition of organocopper compounds to 4- or 5-substituted 2-cyclohexenones proceeds highly selectively via an *anti*-addition pathway, respectively. Thus, at first glance, it seemed difficult to get high diastereoselectivity for the reaction of **2** with organocopper compounds. However, we had previously found that, while the reaction of **1** with the Gillman cuprates R₂CuLi or higher-order cyanocuprates R₂Cu(CN)Li₂ proceeded via an *anti*-addition pathway, the reaction with lower-order cyanocuprates RCu(CN)Li, exceptionally, afforded *syn*-addition products highly selectively owing to the ligating effect of the alkoxy group.^{1a,b} We, therefore, expected that this *syn*-addition tendency by the reaction with RCu(CN)Li might be applicable to the reaction with **2** irrespective of the presence of the *tert*-butyldimethylsilyloxymethyl substituent at the C-4 position, thus affording highly selectively the 1,4-addition product **8** having the structure where the R group introduced is *cis* to the *tert*-butyldimethylsilyloxy group and *trans* to the *tert*-butyldimethylsilyloxymethyl group. As expected, the reaction of RCu(CN)Li where R is a methyl and primary-, secondary-, and tertiary-alkyl group afforded the 1,4-addition product **8** with the anticipated structure almost exclusively. The phenyl derivative, however, proceeded with lower selectivity and lower yield; this observation is in accord with our earlier result observed for the reaction with **1**.^{1b} We also confirmed that the reaction of **2** with R₂Cu(CN)Li₂ provided the mixture of two possible diastereomers, **8** and **9**, in a variable ratio dependent on the R group (Table 1); it is noteworthy, however, that the reaction with Ph₂Cu(CN)Li₂ proceeded with exceptionally high selectivity to afford the corresponding **8** exclusively.⁷

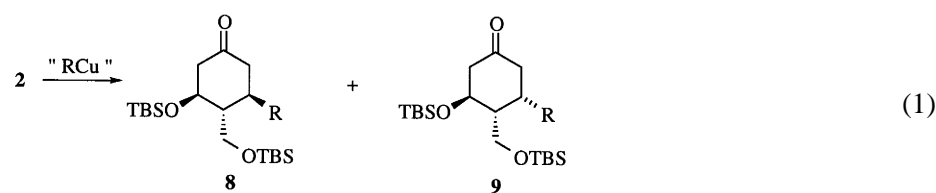
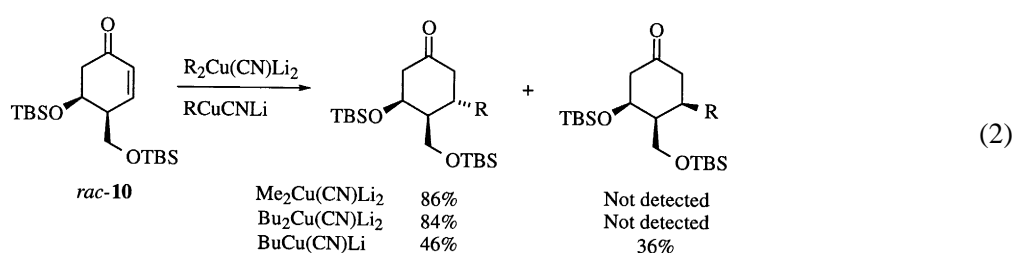


Table 1
1,4-Addition of lower-order and higher-order cyanocuprates onto **2**

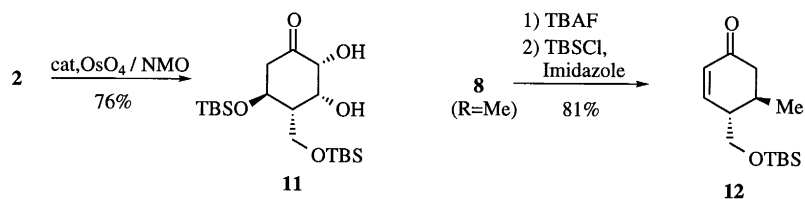
Entry	R	RCu(CN)Li		R ₂ Cu(CN)Li ₂	
		8 : 9 ^a	Yield(%)	8 : 9 ^a	Yield(%)
1	Me	> 95 : < 5	88	3 : 1	93 ^b
2	Bu	> 99 : < 1 ^c	86	58 : 42 ^c	83 ^b
3	<i>sec</i> -Bu	> 95 : < 5	83	5.5 : 1	88 ^b
4	<i>tert</i> -Bu	> 95 : < 5	84	7 : 1	69 ^b
5	Ph	2 : 1	32 ^b	> 95 : < 5	84

^a: NMR measurement unless otherwise noted. <5 : >95 means that the other diastereomer was not detected ^b: NMR yield. ^c: GC measurement.

At this point, we were interested in the diastereoselectivity of the reaction of organocopper compounds with 4-(*tert*-butyldimethylsiloxymethyl)-5-(*tert*-butyldimethylsiloxy)-2-cyclohexenone having *cis* configuration, **10**, i.e., the diastereomer of **2**. We assumed that the reaction with R₂Cu(CN)Li₂ would proceed highly selectively to introduce the R group at the *trans* position to both the *tert*-butyldimethylsiloxymethyl and *tert*-butyldimethylsiloxy groups, while the reaction with RCu(CN)Li might occur with low stereoselectivity. Our assumption turned out to be valid: thus, as shown in Eq. (2), the reaction of *rac*-**10** with Me₂Cu(CN)Li₂ or Bu₂Cu(CN)Li₂ provided the 1,4-addition product with the expected stereochemistry almost exclusively (<5:>95), but the reaction with BuCu(CN)Li afforded a mixture of the two possible adducts in a ratio of 56:44. Thus, it may safely be said that the 1,4-addition reaction of organocopper compounds to 5-(*tert*-butyldimethylsiloxy)-2-cyclohexenone having a substituent at the 4-position can be carried out with excellent selectivity irrespective of the stereochemistry of the enone by using either RCu(CN)Li or R₂Cu(CN)Li₂.



Further synthetic utility of **2** and **8** (obtained from **2** and RCu(CN)Li) in organic syntheses is currently under study in our laboratory. We have so far found that the osmylation of **2** proceeds in a stereoselective way to afford the diol **11** exclusively.⁸ The treatment of **8** (R=Me) with TBAF resulted in β-elimination to give **12**. The relative configuration of **12** was confirmed by converting to *trans*-3-methyl-4-(*tert*-butyldimethylsiloxymethyl)-cyclohexanone⁹ by hydrogenation of the double bond of **12**.



Acknowledgements

G.H. thanks the Japan Society for the Promotion of Science for financial support.

References

1. (a) Hikichi, S.; Hareau, G.; Sato, F. *Tetrahedron Lett.* **1997**, *38*, 8299. (b) Hareau, G.; Hikichi, S.; Sato, F. *Angew. Chem., Int. Ed.* **1998**, *37*, 2099. (c) Hareau, G.; Koiwa, M.; Hikichi, S.; Sato, F. *J. Am. Chem. Soc.* **1999**, *121*, 3640. (d) Hareau, G.; Koiwa, M.; Hanazawa, T.; Sato, F. *Tetrahedron Lett.* **1999**, *40*, 7493, see also (e) Koiwa, M.; Morizono, D.; Hareau, G.; Sato, F. *Tetrahedron Lett.* **1999**, *40*, 4199.
2. Nicolaou, K. C.; Papahatjis, D. P.; Claremon, D. A.; Magolda, R. L.; Dolle, R. E. *J. Org. Chem.* **1985**, *50*, 1440.
3. Kikuchi, Y.; Kurata, H.; Nishiyama, S.; Yamamura, S.; Kato, K. *Tetrahedron Lett.* **1997**, *38*, 4795.
4. Kolb, H. C.; Sharpless, K. B. *Tetrahedron* **1992**, *48*, 10515.
5. (a) Kasatkin, A.; Sato, F. *Tetrahedron Lett.* **1995**, *36*, 6079. (b) Kasatkin, A.; Kobayashi, K.; Okamoto, S.; Sato, F. *Tetrahedron Lett.* **1996**, *37*, 1849. (c) Lee, J.; Kang, C. H.; Cha, J. K. *J. Am. Chem. Soc.* **1996**, *118*, 291.
6. Ito, Y.; Fujii, S.; Nakatsuka, M.; Kawamoto, F.; Saegusa, T. *Org. Synth. Coll. Vol. 6* **1988**, 327.
7. The stereochemistry of **8** (R=Ph) was confirmed by converting to the known 4-hydroxymethyl-5-phenyl-2-cyclohexenone; (Kozmin, S. A.; Rawal, V. H. *J. Org. Chem.* **1999**, *64*, 3039).
8. The absolute configuration was determined by NOE.
9. Rosen, T.; Taschner, M. J.; Thomas, J. A.; Heathcock, C. H. *J. Org. Chem.* **1985**, *50*, 1190.